PHYTOREMEDIATION OF HEAVY METALS CONTAMINATED SOILS

Ankush Sheoran¹* and Sunita Sheoran²

¹Department of Chemistry, Panjab University, Chandigarh ²Department of Soil Science, College of Agriculture, CCS Haryana Agricultural University, Hisar Email: <u>sheoransunita27@gmail.com</u>

Received-06.10.2017, Revised-22.10.2017

Abstract: Soils may be contaminated by the accumulation of heavy metals and metalloids through the emissions from rapidly expanding industrial areas, mine tailings, disposal of high metal wastes, leaded gasoline and paints, application of fertilizers, animal manures, sewage sludge, pesticides, wastewater irrigation etc. Excessive accumulation of heavy metals can have deleterious effects on soil fertility and productivity, disrupts ecosystem functioning and can lead to serious health risks to animals and human beings. Many methods of preventing or removing these pollutants from soils are identified, however, most of these conventional remedial processes are expensive and adversely affect the soil fertility and productivity. Therefore, phytoremediation which uses higher plants to reduce contaminant levels in soil is an eco-friendly and cost effective technology. The objective of this review is to discuss the different mechanisms of phytoremediation, their potentials, limitations, and techniques to enhance the phytoremediation efficiency.

Keywords: Phytoextraction, Hyperaccumulator, Ecofriendly, Cost effective, Chelates, Microbes

REFRENCES

Ali, H., Naseer, M. and Sajad, M.A. (2012). Phytoremediation of heavy metals by *Trifoliumalexandrinum*. International Journal of Environmental Sciences, 2(3):1459-1469.

Banuelos, G., Terry, N., Leduc, D. L., Elizabeth, A.H., Smits, P. and Mackey, B. (2005). Field trials of transgenic Indian mustard plants shows enhanced phytoremediation of selenium contaminated sediment. Environ Science and Technology, 39(6):1771-78.

Basta, N. T., Ryan, J. A. and Chaney, R. L. (2005). Trace element chemistry in residual-treated soil: key concepts and metal bioavailability. Journal of Environmental Quality, 34(1):49–63.

Bianchi, M.J., Masciandaro, G., Giraldi, D., Ceccanti, B. and Iannelli, R. (2008). Enhanced heavy metal phytoextraction from marine dredged sediments comparing conventional chelating agents (citric acid and EDTA) with humic substances. Water Air Soil Pollution, 193:323-333.

Blaylock, M.J. and Huang, J.W. (2000). Phytoextraction of metals. In: Raskin, I.,Ensley, B.D. (Eds.), Phytoremediation of Toxic Metals: Using Plants to Clean upthe Environment. – John Wiley and Sons, New York, pp. 53–70.

Brennan, M. A. and Shelley, M. L. (1999). A model of the uptake, translocation, and accumulation of lead (Pb) by maize for the purpose of phytoextraction. Ecological Engineering, 12:271-297.

Brooks, R.R., Chambers, M.F., Nicks, L.J. and Robinson, B.H. (1998). Phytomining. Trends in Plant and Science, 1: 359-362.

Cao, X., Ma, L.Q. and Shiralipour, A. (2003). Effects of compost and phosphate amendments on arsenic mobility in soils and arsenic uptake by the

hyperaccumulator *Pteris vittata* L. Environmental Pollution, 126:157–167.

Chen T. C. and Hong, A. (1995). Chelating extraction of lead and copper from an authentic contaminated soil using N- (2-acetamido)iminodiacetic acid and S-carboxymethyl-Lcysteine. Journal of Hazardous Materials, 41(2-3):147–160.

Chen, Y.X., Lin, Y.M., He, Y.F., Zhen, S.J., Yu, Y.L., Tian, G.M. and Wong, M.H. (2003). The role of citric acid on the phytoremediation of heavy metal contaminated soil. Chemosphere, 50:807-811.

Clemens, S. (2006). Toxic metal accumulation, responses to exposure and mechanisms of tolerance in plants. Biochimie, 88(11):1707–1719.

Cunningham, S. D. and Ow, D. W. (1996). Promises and prospects of phytoremediation. Plant Physiology, 110(3):715–719.

Cunningham, S. D., Anderson, T. A, Schwab, P. A, and Hsu, F. C. (1996). Phytoremediation of soilscontaminated with organic pollutants. Advances in Agronomy, 56: 55-114.

DeVolder, P. S., Brown, S. L., Hesterberg, D. and Pandya, K. (2003). Metal bioavailability and speciation in a wetland tailings repository amended with biosolids compost, wood ash, and sulfate. Journal of Environmental Quality, 32(3):851–864.

Diels, N., van der Lelie, D. and Bastiaens, L. (2002). New developments in treatment of heavy metal contaminated soils. Reviews in Environmental Science and Biotechnology, 1:75–82.

Duffus, J.H. (2002). Heavy metals—A meaningless term? Pure and Applied Chemistry, 74:793–807.

Farrell, M., Perkins, W. T., Hobbs, P. J., Griffith, G. W. and Jones, D. L. (2010). Migration of heavy metals in soil as influenced by compost amendments. Environmental Pollution, 158(1): 55–64.

*Corresponding Author

Journal of Plant Development Sciences Vol. 9 (10) : 905-915. 2017

Fawzy, E. M. (2008). Soil remediation using in situ immobilization techniques. Chemistry and Ecology, 24(2):147-156.

Garbisu, C. and Alkorta, I. (2001). Phytoextraction: a cost-effective plant-based technology for the removal of metals from the environment. Bioresource Technology, 77(3):229–236.

Garbisu, C. and Alkorta, I. (2001). Phytoextraction: a cost-effective plant-based technology for the removal of metals from the environment. Bioresource Technology, 77: 229-236.

Garbisu, C., Hernandez-Allica, J., Barrutia, O., Alkorta, I. and Becerril, J.M. (2002) Phytoremediation: A technology using green plants to remove contaminants from polluted areas. Reviews on Environmental Health, 17: 75–90.

Ghosh, M. and Singh, S. P. (2005). A review on phytoremediation of heavy metals and utilization of its byproducts. Applied Ecology and Environmental Research, 3(1):1–18.

Giasson, P., Jaouich, A., Gagne, S. and Moutoglis, P. (2005). Arbuscular mycorrhizal fungi involvement in zinc and cadmium speciation change and phytoaccumulation. Remediation Journal, 15:75-81.

Gohre, V. and Paszkowski, U. (2006). Contribution of the arbuscular mycorrhizal symbiosis to heavy metal phytoremediation. Planta, 223:1115-1122.

Gosh, S. (2010). Wetland macrophytes as toxic metal accumulators. International Journal of Environmental Sciences, 1 (4):523-528.

Halim, M., Conte, P. and Piccolo, A. (2003). Potential availability of heavy metals to phytoextraction from contaminated soils induced by exogenous humic substances. Chemosphere, 52:265-275.

Hashimoto, Y., Matsufuru, H., Takaoka, M., Tanida, H. and Sato, T. (2009). Impacts of chemical amendment and plant growth on lead speciation and enzyme activities in a shooting range soil: an X-ray absorption fine structure investigation. Journal of Environmental Quality, 38(4):1420–1428.

Huang, J.W., Chen, J., Berti, W. R. and Cunningham, S.D. (1997). Phytoremediation of lead-contaminated soils: role of synthetic chelates in lead phytoextraction. Environmental Science and Technology, 31:800-805.

Ishikawa, S., Noriharu, A.E., Masaharu, M.I. and Tadao, W. (2006). Is *Brassica juncea* a suitable plant for phytoremediation of cadmium in soils with moderately low cadmium contamination.Possibility of using other plant species for Cd-phytoextraction. Soil Science and Plant Nutrition, 52:32.

Jadia, C. D. and Fulekar, M. H. (2009). Phytoremediation of heavy metals: recent techniques. African Journal of Biotechnology, 8(6):921–928.

Koppolu, L. and Clements, L.D. (2003). Pyrolysis as a technique for separating heavy metals from hyperaccumulators. Part I: Preparation of synthetic

hyperaccumulator biomass. Biomass and Bioenergy, 24: 69–79.

Labanowski, J., Monna, F. and Bermond A. (2008). Kinetic extractions to assess mobilization of Zn, Pb, Cu, and Cd in a metal-contaminated soil: EDTA vs. citrate. Environmental Pollution, 152(3):693–701.

Ling, W., Shen, Q., Gao, Y., Gu, X. and Yang, Z. (2007). Use of bentonite to control the release of copper from contaminated soils. Australian Journal of Soil Research, 45(8):618–623.

MacCarthy, P. (2001). The principles of humic substances. Soil Science, 166:738–751

Mandal, A., Purakayastha, T. J., Patra A. K. and Sanyal, S. K. (2012). Phytoremediation of Arsenic Contaminated Soil by Pteris Vittata L. II. Effect on Arsenic Uptake and Rice Yield. International Journal of Phytoremediation, 14(6):621-628.

Martin, T. A. and Ruby, M. V. (2004). Review of in situ remediation technologies for lead, zinc and cadmium in soil. Remediation, 14(3):35–53.

McGrath, S.P., Zhao, F.J. and Lombi, E. (2002). Phytoremediation of metals, metalloids, and radionuclides. Advances in Agronomy, 75: 1–56.

Meda, A.R., Scheuermann, E.B., Prechsl, U.E., Erenoglu, B., Schaaf, G., Hayen, H., Weber, G. and von Wiren, N. (2007). Iron acquisition by phytosiderophores contributes to cadmium tolerance. Plant Physiology, 143:1761-1773.

Mendez, M. O. and Maier, R. M. (2008). Phytostabilization of mine tailings in arid and semiarid environments-an emerging remediation technology. Environmental Health Perspectives, 116:278-283.

Panwar, B.S., Kadar, I., Biro, B., Rajkai-Vegh, K., Ragayi, P., Rekasi, M. and Marton. L. (2011). Phytoremediation: enhanced cadmium (cd) accumulation by organic manuring, EDTA and microbial inoculants (*Azotobactersp., Pseudomonas* sp.) in Indian mustard (*Brassica junceaL.*). Acta Agronomica Hungarica, 59(2):101–107.

Peng, K.J., Luo, C.L., Chen, Y.H., Wang, G.P., Li, X.D. and Shen, Z.G. (2009). Cadmium and other metal uptake by *Lobelia chinensis* and *Solanum nigrum* from contaminated soils. Bulletin of Environmental Contamination and Toxicology, 83:260-264.

Quartacci, M.F., Argilla, A., Baker, A.J.M. and Navari-Izzo, F. (2006). Phytoextraction of metals from a multiply contaminated soil by Indian mustard. Chemosphere, 63:918-925.

Ramprakash, Kumari, S., Sangwan, A., Rajpaul and Kumar, S. (2013). Phytoextraction of chromium from contaminated soil by *Brassica juncea* as influenced by chelating agents. Asian Journal of Chemistry, 25(10):5357-5359.

Raskin, I. and Ensley, B. D. (2000). Recent developments for in situ treatment of metal contaminated soils. In: *Phytoremediation of Toxic*

Metals: Using Plants to Clean Up the Environment. JohnWiley & Sons Inc., New York.

Raskin, I., Smith R.D. and Salt, D.E. (1997): Phytoremediation of metals: Using plants to remove pollutants from the environment. Current Opinion in Biotechnology, 8(2): 221-226.

Raven, P. H., Berg, L. R. and Johnson, G. B. (1998). *Environment*, Saunders College Publishing, 2nd edition. New York, NY, USA.

Reisinger, S., Schiavon, M., Terry, N. and Pilon-Smits, E.A.H. (2008).cHeavy metal tolerance and accumulation in Indian mustard (Brassica juncea L.) expressing bacterial gamma-glutamyl cysteine synthetase or glutathione synthetase. International Journal of Phytoremediation, 10:440-454.

Rizzi, L., Petruzelli, G., Poggio, G. and VignaGuidi, G. (2004). Soil physical changes and plant availability of Zn and Pb in a treatability test of phytostabilization. Chemosphere, 57:1039-1046.

Ross, A., Suther and Filip, M.G. (2003). Tack, Fractionation of Cu, Pb and Zn in certified reference soils SRM 2710 and SRM 2711 using the optimized BCR sequential extraction procedure. Advances in Environmental Research, 8:37–50.

Rramprakash, Singh, S.K., Singh, A. (2009). Effect of chelating agents on distribution of lead in soil after harvest of Brassica juncea. Environment and Ecology, 27 (3):981-984.

Salt, D.E., Blaylock, M., Kumar, P.B.A.N., Dushenkov, V., Ensley, B.D., Chet, I. and Raskin, I. (1995). Phytoremediation: a novelstrategy for the removal of toxic metals from the environment using plants. Biotechnology, 13: 468–475.

Salt, D.E., Smith, R.D. and Raskin, I. (1998). Phytoremediation. Annual Review of Plant Physiology and Plant Molecular Biology, 49: 643– 668.

Sekara, A., Poniedzialek, M., Ciura, J. and Jedrszczyk, E. (2005). Cadmium and lead accumulation and distribution in the organs of nine crops: Implications for phytoremediation. Polish Journal of Environmental Studies, 14:509-516.

Singh, S.K., Ramprakash, Kumari, S. and Duhan, B.S. (2013). Phytoextraction of Ni from contaminated soil by *Brassica juncea* as influenced by chelating agents. Annals of Biology 29(1):15-18.

Smith, A.H., Lopipero, P.A., Bates, M.N. and Steinmaus, C.M. (2002). Public health arsenic epidemiology and drinking water standards. Science, 296:2145–2146.

Sumner, M. E. (2000). Beneficial use of effluents, wastes, and biosolids, Communications in Soil Science and Plant Analysis, 31(11–14):1701–1715.

Tandy, S., Bossart, K., Mueller, R., Ritschel, J., Hauser, L., Schulin, R. and Nowack, B. (2004). Extraction of heavy metals from soils using biodegradable chelating agents. Environmental Science and Technology, 38:937-944.

Trap, S., Kohler, A., Larsen, L. C., Zambrano, K. C. and Karlson, U. (2005). Phytotoxicity of fresh and weathered diesel and gasoline to willow and poplar trees. Journal of Soils and Sediments, 1: 71-76.

United States Environmental Protection Agency (**USEPA**). (2000). *Introduction to Phytoremediation*. EPA 600/R-99/107, U.S. Environmental Protection Agency, Office of Research and Development, Cincinnati, OH.

Vamerali, T., Bandiera, M., Coletto, L., Zanetti, F., Dickinson, N.M. and Mosca, G. (2009). Phytoremediation trials on metal and arseniccontaminated pyrite wastes (Torviscosa, Italy). Environmental Pollution, 157:887-894.

Wang, L. Q., Luo, L., Ma, Y. B., Wei, D. P. and Hua, L. (2009). In situ immobilization remediation of heavy metals-contaminated soils: a review. Chinese Journal of Applied Ecology, 20(5):1214–1222.

Wenger, K., Gupta, S.K., Furrer, G. and Schulin, R. (2003). The role of nitrilotriacetate in copper uptake by tobacco. Journal of Environmental Quality, 32:1669-1676.

Wright, R.T. (2007). Environmental Science: Toward a Sustainable Future. 9th Ed. Prentice Hall of India, New Delhi.

Wuana, R. A., Okieimen, F. E. and Ikyereve, R. E. (2008). Removal of lead and copper fromcontaminated kaolin and bulk clay soils using acids and chelating agents. Journal of Chemical Societyof Nigeria, 33(1):213–219.

Yang, Q., Tu, S., Wang, G., Liao, X. and Yan, X. (2012). Effectiveness of Applying Arsenate Reducing Bacteria to Enhance Arsenic Removal From Polluted Soils by Pteris Vittata L. International Journal of Phytoremediation, 14(1): 89-99.

Yoon, J., Cao, X. and Zhou, O. (2006). Accumulation of Pb, Cu and Zn in native plants growing on a contaminated Florida site. Science Total Environment, 368:456-464.